

## HIGH POLARIZATION DOPANTS FOR FERROELECTRIC LIQUID CRYSTAL COMPOSITIONS

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### BACKGROUND OF THE INVENTION

The present invention relates generally to liquid crystal compounds and compositions and to devices employing liquid crystal compositions. The invention more specifically relates to chiral nonracemic compounds useful as dopants in ferroelectric liquid crystal compositions to impart high polarization and fast switching speed. The dopants combine a rod-like mesogenic core with a chiral nonracemic tail and an achiral tail.

Several types of smectic LC materials have been investigated for rapid switching, view-angle enhancement and higher contrast, including surface-stabilized ferroelectric LCs (SSFLCs), deformed helix ferroelectric LCs (DHFLCs) and antiferroelectric LCs (AFLCs). Recently, smectic materials exhibiting thresholdless or more properly V-shaped switching LCs (VLCs) have been described (Inui, S. et al. (1996) J. Mater. Chem. 6(4):671-673; Seomun, S.S. et al. (1997) Jpn. J. Appl. Phys. 36:3580-3590).

Liquid crystal compositions exhibit one or more LC phases. LC compositions may be composed of one or more components. Components of LC compositions may exhibit liquid crystal phases, have latent liquid crystal phases or be compatible with (not suppress) liquid crystal phases in the liquid crystal composition.

To achieve full color imaging in ferroelectric liquid crystal (FLC) displays, the FLC material is

required to have fast switching speed. It is known in the art that switching speed is proportional to the viscosity of and inversely proportional to the spontaneous polarization (Ps) of the FLC material. It is also known that switching speed can be improved by doping a small percentage of high Ps compounds into an achiral smectic C host.

Chiral FLC compounds and components of LC mixtures generally consist of a rigid mesogenic core having one or more directly or indirectly linked alicyclic or aromatic rings (which may be fused aromatic rings) and achiral and chiral tails distributed on either side of the mesogenic core, e.g.:



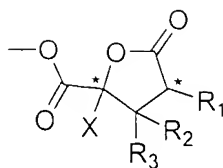
Large Ps can be achieved by modifying one of these three building blocks, but among them, developing new chiral tails with large dipoles which can be oriented in the smectic C phase of a host FLC, is the most efficient way to increase the Ps of the LC composition.

#### SUMMARY OF THE INVENTION

The invention relates to chiral nonracemic liquid crystal compounds which are useful as components in liquid crystal compositions to impart high polarization to the mixture. The materials of this invention can be combined with liquid crystal host materials to impart improved properties to mixtures.

Chiral nonracemic compounds of this invention can function as additives or dopants in host materials to impart chirality into an LC material.

Most generally the invention provides mixtures containing one or more chiral nonracemic dopants which have a chiral nonracemic tail that is an optionally substituted  $\alpha$ -ester  $\gamma$ -lactone of the formula:



Formula I

where \* indicates a chiral carbon;  $R_1$  is a straight-chain or branched alkyl or alkenyl group wherein one or more non-neighboring carbon atoms can be replaced with an oxygen atom and wherein one or more carbons can be substituted with one or more halogens;  $R_2$  and  $R_3$ , independently of one another, can be hydrogen, a halogen or a lower alkyl or alkenyl group; and X is hydrogen or a lower alkyl group. LC compounds having a chiral nonracemic tail of this general formula have large dipoles (and large spontaneous polarization ( $P_s$ )) and when doped into hosts such as achiral smectic C FLC hosts, lead to improved switching speeds of the FLC materials.

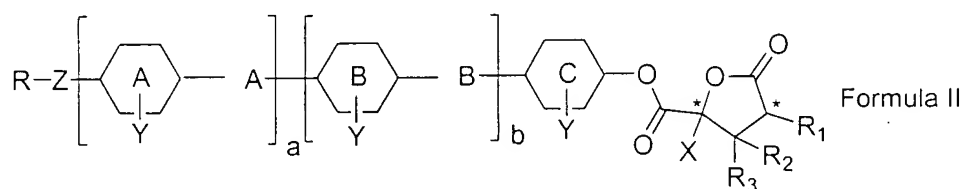
This chiral nonracemic tail imparts high polarization to liquid crystal compositions. Liquid crystal compounds having a chiral non-racemic  $\gamma$ -lactone ester tail bonded to a mesogenic core impart high polarization to liquid crystal compositions containing them. The chiral non-racemic  $\gamma$ -lactone ester tail can be combined with any known mesogenic core, particularly, rod-like, linear cores.

Typically mesogenic cores include one to about three aromatic or alicyclic rings. Aromatic rings include both 5- and 6-member aromatic rings and fused aromatic rings, such as naphthalene rings. One or two carbons in a 5- or 6- member aromatic ring can be replaced with a heteroatom, such a O, N or S. Alicyclic rings include cyclohexane and cyclohexene rings wherein one of the carbons atoms of the ring can be replaced with a Heteroatom or in which one of the  $CH_2$  of the ring can be replaced with  $C=O$ . The carbons of the alicyclic ring or the aromatic ring can be substituted with lower alkyl groups, halides, CN groups or nitro groups. Mesogenic cores also include a dioxane ring, an optionally substituted phenyl ring, an optionally substituted cyclohexyl ring and an optionally substituted cyclohexenyl ring in which one or two carbons may be optionally substituted with the same or different heteroatoms, such as O, N or S.

Mesogenic cores can comprise one aromatic ring, two aromatic rings or three aromatic rings. Mesogenic cores can comprise two or three cyclohexane rings, cyclohexene rings or a combination of both. Mesogenic cores can contain a combination of aromatic rings and cyclohexane or cyclohexene rings. The aromatic or alicyclic rings of a mesogenic core can comprise linking groups between the rings. Linking groups can include a single bond, a  $-\text{CH}_2\text{-CH}_2-$ ,  $-\text{CH=CH-}$ ,  $-\text{C}\equiv\text{C-}$ , O, S, CO, COO,  $-\text{CH}_2\text{-O-}$ .

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More specifically, the invention relates to chiral nonracemic compounds of general formula:



where \*,  $R_1$ ,  $R_2$ ,  $R_3$ , and X are as described for Formula I;

R is selected from the group consisting of:

(1) an achiral straight chain or branched silane or siloxane having one or more silicon atoms and which may be substituted with one or more halogens;

(2) an achiral linear or branched perfluorinated or partially fluorinated alkyl group ( $R^F$ );

✓ (3) an achiral linear, cyclic or branched perfluorinated or partially fluorinated ether group;

(4) an achiral linear or branched ether having one or more oxygen atoms and which may be substituted with one or more halogens;

(5) an achiral alkyl, alkenyl or alkynyl group which may be substituted with one or more halogens; and

(6) a straight chain or branched thioether having one or more sulfur atoms and which may be substituted with one or more halogens.

and where:

Z is a linker selected from the group consisting of O, CO, OOC, COO, S or a single bond;

core rings A, B and C and linker moieties A and B together represent a mesogenic core, and can be aromatic or alicyclic; if aromatic, one or two ring carbons can be replaced with a nitrogen; or if alicyclic, rings can contain 3-10 carbon atoms and optionally can contain a double bond, wherein one or two CH<sub>2</sub> of the alicyclic ring can be replaced with a nitrogen, sulfur, or oxygen atom, or a C=O group, each ring can be a fused ring system;

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Y represents up to four substituents on a given ring when the ring is aromatic and up to 20 substituents when the ring is alicyclic, where substituents are selected from a halide, CN, NO<sub>2</sub>, alkyl or alkoxy groups;

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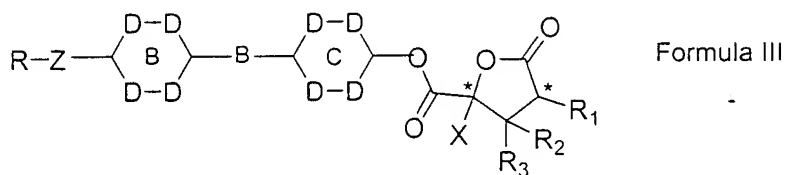
Linkers A and B, independently, are selected from the group consisting of a single bond, -COO-, -OOC--CH<sub>2</sub>-CH<sub>2</sub>-, -OCH<sub>2</sub>-, -CH<sub>2</sub>-O-, -CH=CH- (cis or trans); -C≡C-, and -CH=CH-CH=CH- (cis or trans);

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and where a and b are integers that are 0 or 1 to indicate the absence or presence of the groups in brackets and where a+b is 1 or 2.

In a specific embodiment, compounds of this invention have the formula:

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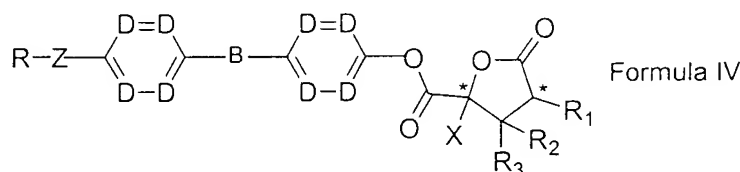


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where R, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, X, linker B, and rings B and C are as defined above and each D, independent of other D's, can be CH, CH<sub>2</sub>, C-CH<sub>3</sub>, CH-CH<sub>3</sub>, a nitrogen atom, CY or CHY, where Y is a CN, NO<sub>2</sub>, an alkyl, a perhaloalkyl (e.g., perfluoroalkyl), or a halide, (e.g., fluorine). Rings B and C can be alicyclic or aromatic and B and C that are aromatic can be fused ring systems, such as naphthalene. One of rings B or C can also be a fused ring system that is partially aromatic, such as a dehydronaphthalene ring

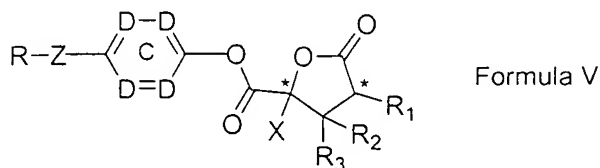
system. In particular embodiments, both of rings B and C are aromatic, or one of rings B or C is aromatic and the other of rings B or C is alicyclic, particularly a cyclohexane or cyclohexene ring. In preferred embodiments: (1) all D's are CH; (2) one or two D's are N and the remaining D's are CH; (3) one or two D's are CF and the remaining D's are CH; (4) one or two D's are N, one or two D's are CF and the remaining D's are CH; (3) all D's on one ring are CH<sub>2</sub> and one, two or three D's on the other ring can be N or CF; (4) all D's on one ring are CH<sub>2</sub> and all D's on the other ring are CH; (5) one or two D's on one ring are CF and one D on the other ring is C-CH<sub>3</sub>.

In a further specific embodiment, compounds of this invention have the formula:



where R, Z, B, X, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are as defined above, and where each D independent of other D's can be CH or CY as defined above or a nitrogen atom and all other variables are as defined above. In preferred compounds of Formula IV, all D's are CH or one or two D's can be CF or nitrogen with the remaining D's being CH. In preferred embodiments, the core is a phenylpyrimidine, a phenylpyridine, phenylbenzoate, or biphenyl.

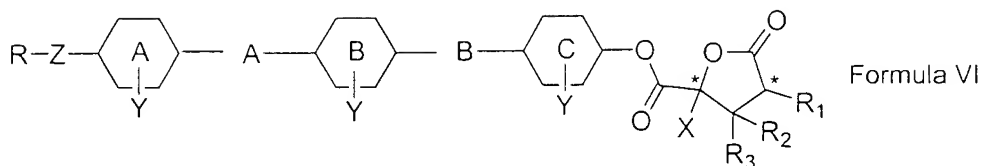
In another specific embodiment, compounds of this invention can have the formula:



where the variables R, Z, C, X, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are as defined above, and where each D, independent of other D's, can be CH, CY or a nitrogen. In preferred compounds of Formula V,

all D's are CH or one or two D's can be CF or nitrogen with the remaining D's being CH.

In yet another specific embodiment, compounds of this invention can have the formula:



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where the variables are as defined in Formula II above. In particular embodiments of Formula VI the core can be (1) an optionally substituted terphenyl, where the preferred substitution is one or two F's; (2) a core in which A or C is a cyclohexane or cyclohexene ring and the remaining rings are aromatic; (3) a core in which A or C is a cyclohexane or cyclohexene and the remaining rings are selected from phenyl rings, phenyl rings substituted with one or two F's, pyrimidine rings or pyridine rings; or (4) a core in which A or C is a cyclohexane or cyclohexene and the remaining two rings represent a phenylpyrimidine, a phenylpyridine, a phenyl benzoate or a biphenyl.

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Compounds of Formulas II - VI are useful in the preparation of LC and FLC compositions which in turn are useful in various optical device applications.

Particular subsets of compounds of Formula II - VI include those in which:

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R is an achiral alkyl, alkenyl or alkynyl group having from 3 to about 20 carbon atoms in which one or more of the non-neighboring carbons can be replaced with an oxygen, or in which one or more of the carbons is substituted with one of more halogens;

R is R<sup>F</sup> where R<sup>F</sup> is an achiral linear or branched perfluorinated or partially fluorinated alkyl group;

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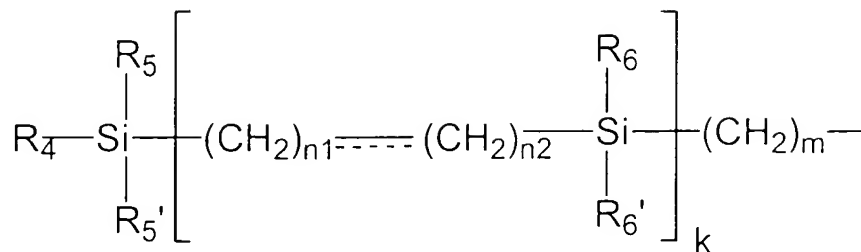
R<sup>F</sup> has the formula: C<sub>n</sub>F<sub>2n+1</sub>C<sub>m</sub>H<sub>2m</sub> wherein n is an integer ranging from 1 to about 10 and m is an integer ranging from 1 to about 10;

R<sup>F</sup> has the formula: C<sub>n</sub>F<sub>2n+1</sub>C<sub>m</sub>H<sub>2m</sub> wherein n is an integer ranging from 1 to about 20 and m is an

integer ranging from 0 to about 20;

$R^F$  is  $C_4F_9C_4H_8$ ;

R is an achiral silane:



where:

$R^4$  is a straight chain or branched alkyl or alkenyl group having one or more carbon atoms and  $R_5$ ,  $R_5'$ ,  $R_6$  and  $R_6'$ , independently of one another, are alkyl groups having from 1-6 carbon atoms;

$n_1$  and  $m$  are integers from 1 to about 20;

$n_2$  can be zero or an integer from 1 to 20 where the dashed line indicates a possible double or triple bond;

$k$  is 0 or an integer from 1 to 10;

X is H;

X is a lower alkyl group;

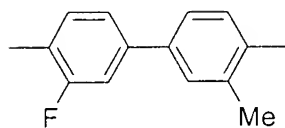
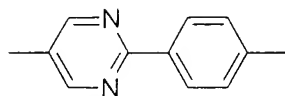
X is  $CH_3$

Z is O;

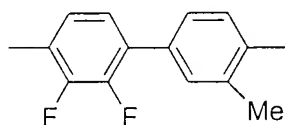
Z is a single bond;

the mesogenic core is





OR



As used herein the term alkyl refers generally to straight-chain and branched alkyl groups. Alkyl groups can include lower alkyl groups (those having from 1-6 carbon atoms) and higher alkyl groups (those having about 7 or more carbon atoms), unless otherwise noted. The term alkoxy group refers to groups having a terminal oxygen atom (e.g., -O-alkyl). For example, alkoxy tail groups are attached to the core via the terminal oxygen. The alkyl portion of an alkoxy group includes straight-chain and branched alkyl groups and unless otherwise noted includes lower alkyl and higher alkyl groups. Alkyl groups, including those of alkoxy group, typically have less than 20 carbons and preferably, dependent upon the specific structure, have 12 or fewer carbon atoms. In compounds where alkyl or alkoxy tail groups are specified, preferred alkyl groups have from 5 to 12 carbon atoms and more preferred alkyl groups have 6 to 10 carbon atoms.

As used herein the term alkene refers generally to any group containing one or more double bonds. The alkene tails of this invention as specified in Formulas I-VI contain a single double bond. Alkene tails include alkene-oxy tails, i.e., -O-alkene, in which the alkene group has a terminal oxygen atom which forms the bond to the core. In general the double bond of the alkene tail can be positioned anywhere in the chain, but preferably is located 2 or more carbons from the end of the tail attached to the



core. The alkene may contain an omega double bond, but the double bond is more preferably located away from the ends of the tail. The double bond may be in the cis or trans configuration.

The term alicyclic generally refers to alkyl or alkene groups that contain a cyclic portion. An alicyclic group can be a saturated ring or unsaturated ring, such as a cyclohexane or cyclohexene ring.

5 Alicyclic rings can contain one or more (typically one) heteroatoms, e.g., O, N or S, in place of ring CH<sub>2</sub> groups. Further, one or more (typically one) ring CH<sub>2</sub> groups can be replaced with C=O groups. Alicyclic groups of the cores of this invention are optionally substituted (unless otherwise noted). Preferred substituents include lower alkyl groups, lower alkene groups, halogens, CN and NO<sub>2</sub> groups. Preferred  
10 halogen substituents are fluorines. In general, all but two aromatic ring positions (e.g., the positions for linkages to tails or to other core rings) can carry non-hydrogen substituents. However, more typically one or two ring positions (in addition to the linkages to the tails or other cores) can be substituted.

The term aromatic generally refers to a group containing at least one aromatic ring, e.g., a phenyl ring. Aromatic rings typically are five or six-member aromatic rings. Aromatic rings can also include fused  
15 aromatic rings, such as naphthalene or dehydronaphthalene rings (see Scheme 1). Scheme 1 shows some mesogenic cores useful in compounds of the invention. An aromatic ring can contain one or more (typically one or two) heteroatoms, e.g., O, N or S. Aromatic groups of the cores of this invention are optionally substituted (unless otherwise noted). Preferred substituents include lower alkyl groups, lower alkene groups, halogens, CN and NO<sub>2</sub> groups. Preferred halogen substituents are fluorines. In general,  
20 all but two positions on the ring can be substituted (e.g., the positions for linkages to tails or to other core rings). However, typically one to four positions of the ring can be substituted and more typically one or two ring positions (in addition to the linkages to the tails or other cores) can be substituted. Preferred substituted aromatic rings have one position substituted with a lower alkyl or alkene group, a CN group or a NO<sub>2</sub> group. Additionally preferred substituted aromatic rings have one or two positions substituted with  
25 one or two halogens, and the preferred halogen is fluorine.



The invention provides LC compositions comprising one or more of the compounds of this invention as described above. LC compositions of the invention include those that contain sufficient amounts of one or more of the compounds of this invention to have a substantial effect upon the physical or optical properties of the LC composition in which they are combined or to which they are added. A substantial effect upon the physical or optical properties of the LC compositions includes, among others, a measurable change in a LC property of the composition, for example, switching speed. LC compositions of this invention include those that contain from about 1% to 100% by weight of one or more compounds of this invention. LC compositions of this invention include those that contain 1% or more by weight of one or more of the compounds of this invention. Preferably, LC compositions of this invention contain 1% to 50% by weight of one or more compounds of this invention. More preferably, LC compositions of this invention contain 1% to 15% by weight of one or more compounds of this invention.

LC compositions of this invention include those that are ferroelectric liquid crystal compositions, particularly those that exhibit smectic phases, and more particularly those that exhibit a smectic C phase. LC compositions of this invention exhibit smectic phases from about -10°C to 70°C.

Any FLC host can be used as known in the art as long as the host does not interfere with the desired effect. One specific host that is used in the examples shown herein is MX6111, shown in Table 2.

LC and FLC compositions of this invention are useful in the preparation of optical devices, particularly for optical switching devices and displays. It is known to those of ordinary skill in the art how to make LC and FLC cells and devices that utilize the compositions of this invention. In particular, methods and techniques are known and available in the art for alignment of LC and FLC compositions between substrate layers to form optical elements that exhibit true bistable, near bistable, or tristable state switching or optical elements that exhibit analog behavior. Various methods and techniques for constructing LC and FLC cells and for use of such cells are known in the art and can be readily adapted



for use with compositions of this invention. The compositions of this invention are particularly well suited for providing devices that can operate (in a smectic C phase, for example) over a broad temperature range.

## DETAILED DESCRIPTION OF THE INVENTION

5 The invention relates generally to compounds that are useful as components in LC compositions. LC compositions typically contain a plurality of components, some of which exhibit LC phases alone or in combination. More preferred LC compositions are those which exhibit a desired LC phase over a temperature range that facilitates practical application of the composition in an optical device. For example, LC materials exhibiting a smectic C range around normal room temperature can  
10 be employed in device applications. Preferred LC materials will exhibit the desired LC phase over a broad, useful temperature range which facilitates device stability.

Preferred LC materials will exhibit a desired LC phase with temperature range that minimizes or avoids crystallization of components of the LC composition during operation or storage of an optical  
15 device. Compounds of this invention can improve (broaden or shift) the temperature range of desired LC phases in LC compositions to which they are added. In particular, compounds of this invention can be employed to broaden or shift the temperature range of smectic C phases of LC compositions. The compounds may also be added to lower the temperature at which crystallization of an LC composition occurs to improve storage lifetime of an LC device containing the LC composition. Improvement is  
20 assessed as lowering of the melting point of the compositions and/or as lowering of the freezing point of the mixture. A significant improvement in LC stability can be obtained with even a 2°C lowering of melting point, if that lowering is obtained without a significant negative effect on other LC properties.

## Examples

25 Table I shows the results for spontaneous polarization as measured for some compounds of this invention. Table 3 shows comparative results for lactone ether compounds. It is seen that compounds of

the invention with lactone ester tails have higher polarization than lactone ether compounds. In all cases, the compounds were doped at 5% by weight in host MX 6111 shown in Table 2. Schemes 4 and 5 illustrates a number of compounds that can be combined with the chiral nonracemic compounds of this invention to provided useful mixtures. Compounds illustrated therein can be prepared by methods that are well known in the art from readily available starting materials. Methods that are useful in the preparation of various LC compounds and FLC compounds are provided, for example in U.S. patents: 5,051,506; 5,061,814; 5,130,048; 5,167,855; 5,178,791; 5,178,793; 5,180,520; 5,271,864; 5,278,680; 5,380,460; 5,422,037; 5,453,218; 5,457,235; 5,539,555; 5,543,078; 5,585,036; 5,626,792; 5,637,256; 5,658,493; 5,753,139; 5,866,036; and 6,139,771. Each of which is incorporated by reference herein for synthetic methods applicable to the synthesis of compounds of this invention. The listed patents along with U.S. patents 5,168,381 and 5,596,434 also provide detail of how LC and FLC compositions of this invention can be applied for the production of LC cells and optical devices.

Concurrently filed U.S. patent applications Attorney Docket Nos. 106-00, 86-00 and 85-00 all provide description of LC components and methods of synthesis of those components that can be combined with the chiral nonracemic compounds of this invention to provide useful FLC compositions.

In addition, chiral racemic compounds or corresponding achiral compounds of this invention can be employed as additional compatible components of FLC compositions of this invention.

Scheme 2 lists an example synthesis to prepare the chiral tails of the invention. The variables in the scheme are as described here. Other synthesis methods are well known to one of ordinary skill in the art.

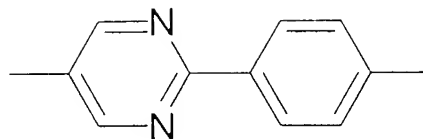
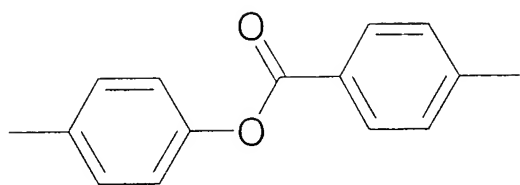
Scheme 3 shows some of the more preferred compounds of the invention.

Although the description above contains many specificities, these are not intended to limit the

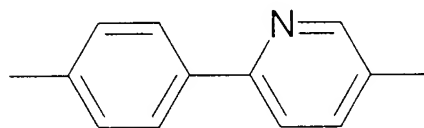
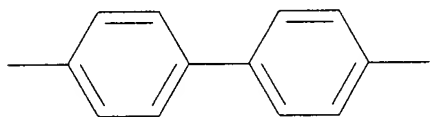
scope of the invention, but merely to provide illustrations of some of the presently - preferred  
embodiments of the invention. For example, mesogenic cores other than those specifically, illustrated  
herein can be used in the compounds of the invention. Also, compounds other than those specifically  
described may be used, as known in the art to form mixtures with the compounds of the invention.  
Thus, the scope of the invention should be determined by the appended claims and their legal  
5 equivalents. All references cited herein are hereby incorporated by reference to the extent not  
inconsistent herewith.

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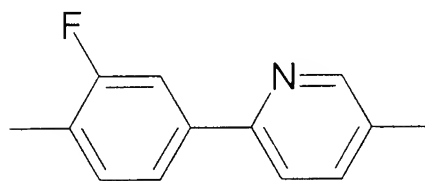
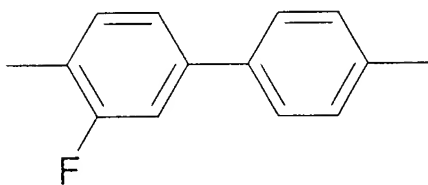
Scheme1 (page 1)



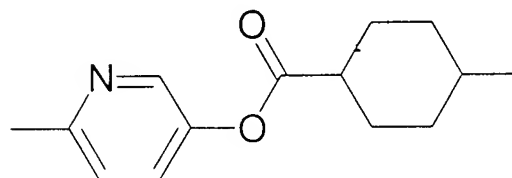
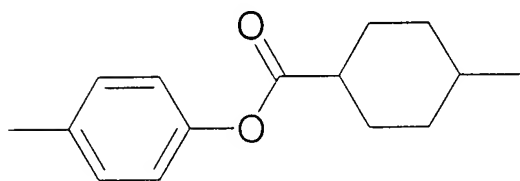
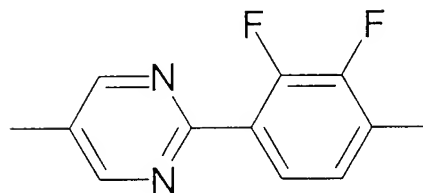
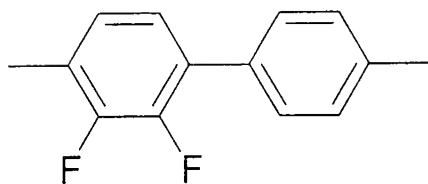
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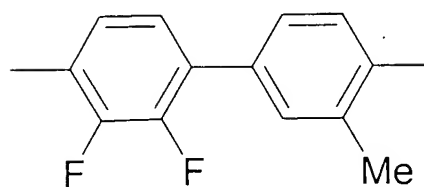
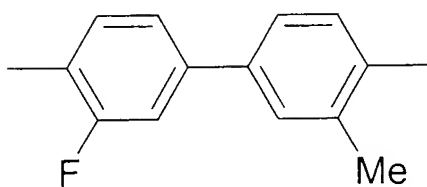
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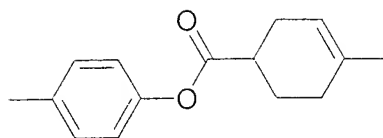
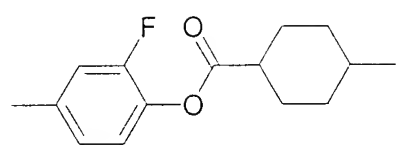
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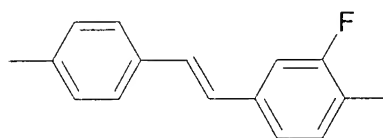
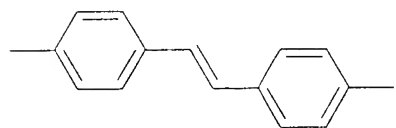
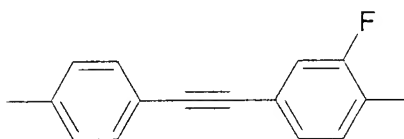
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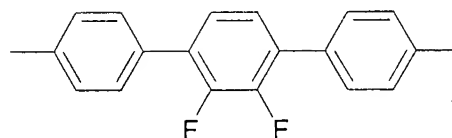
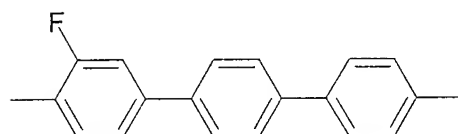
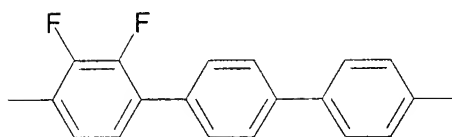
Scheme 1 (page 2)



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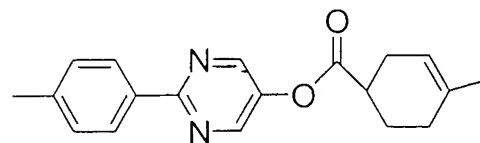
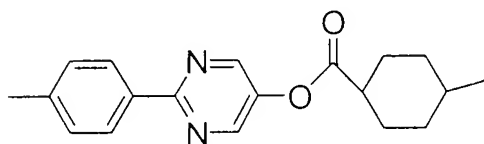
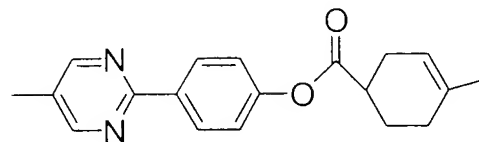
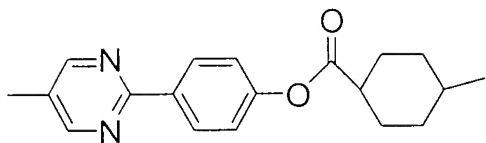
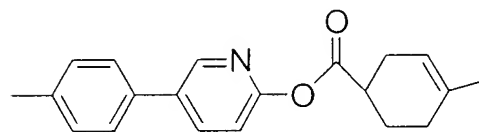
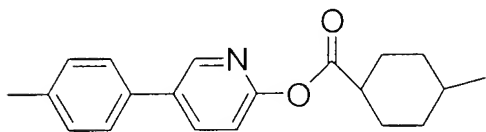
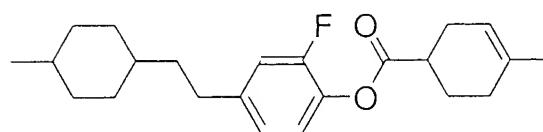
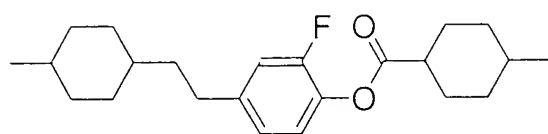
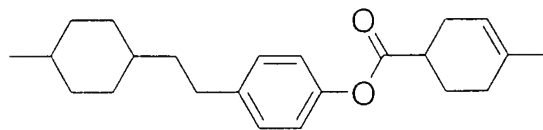
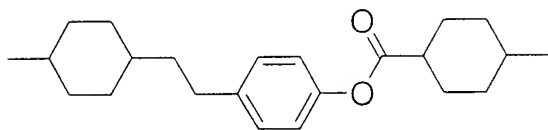
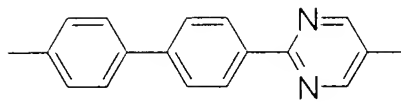
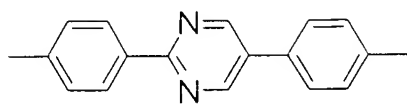


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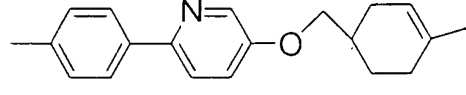
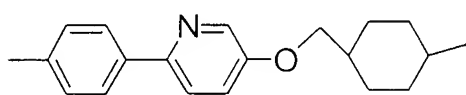
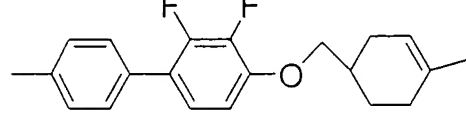
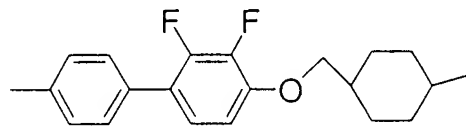
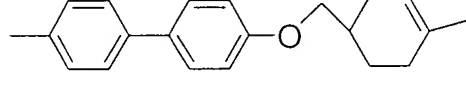
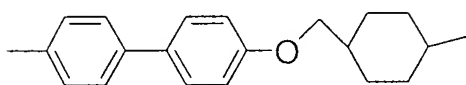
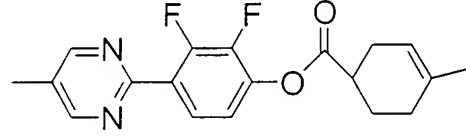
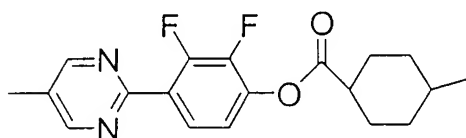
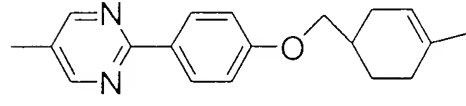
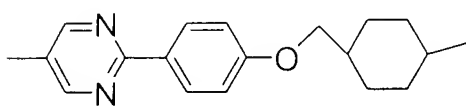
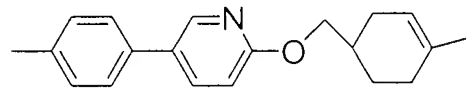
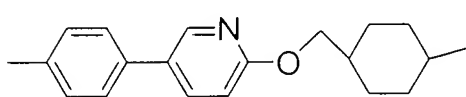
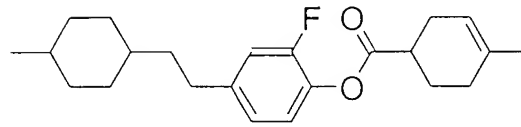
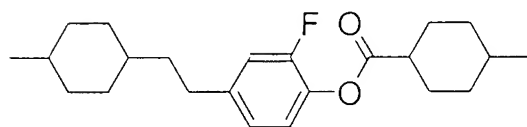
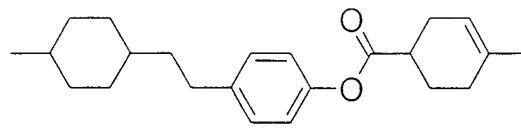
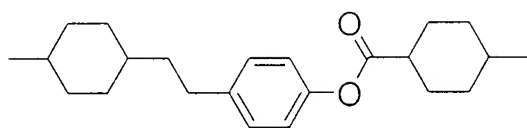
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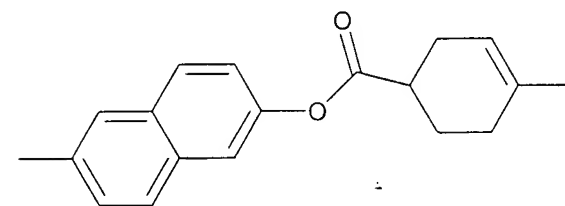
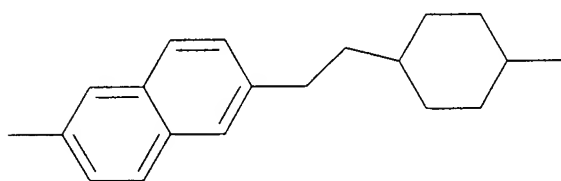
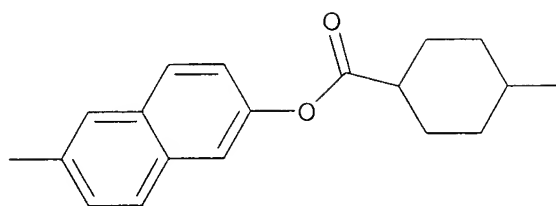
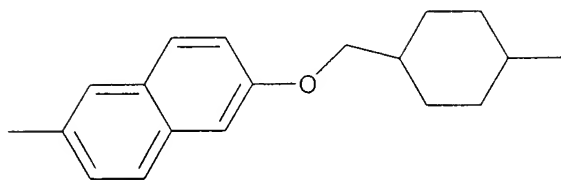
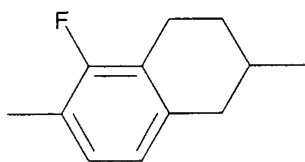
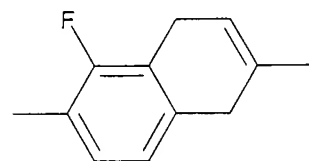
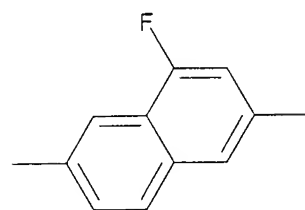
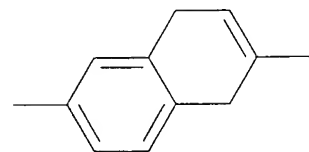
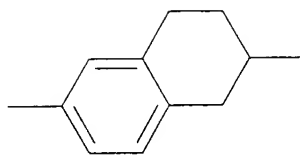
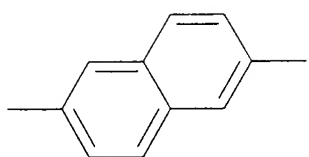
Scheme 1 (page 3)

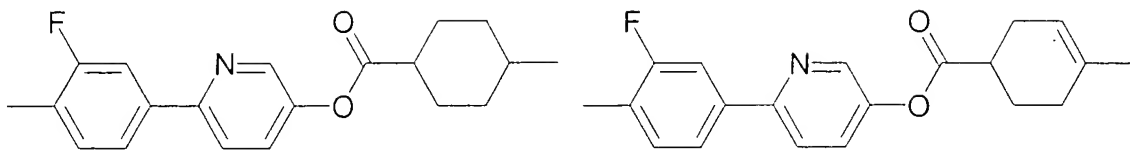


Scheme 1 (page 4)

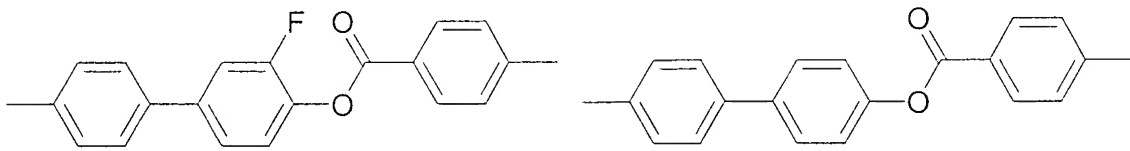


Scheme 1 (page 5)

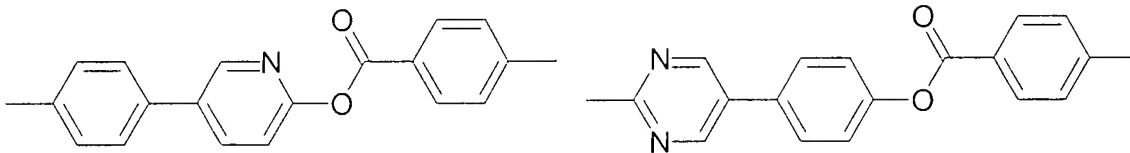




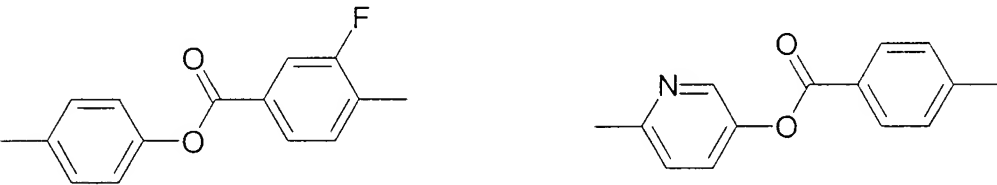
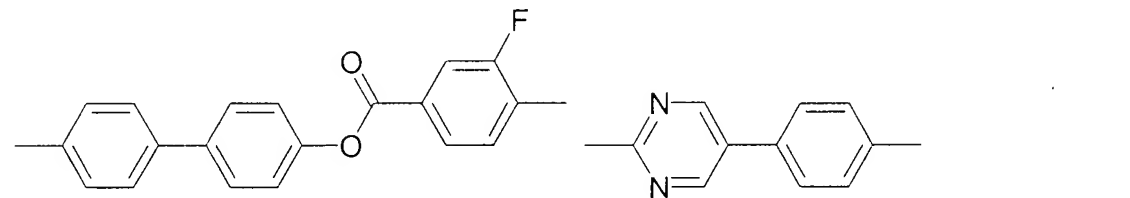
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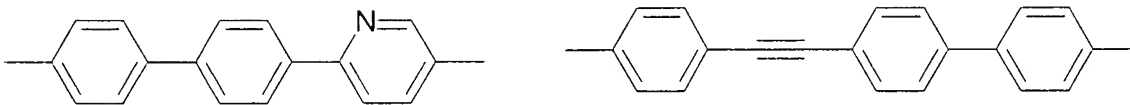
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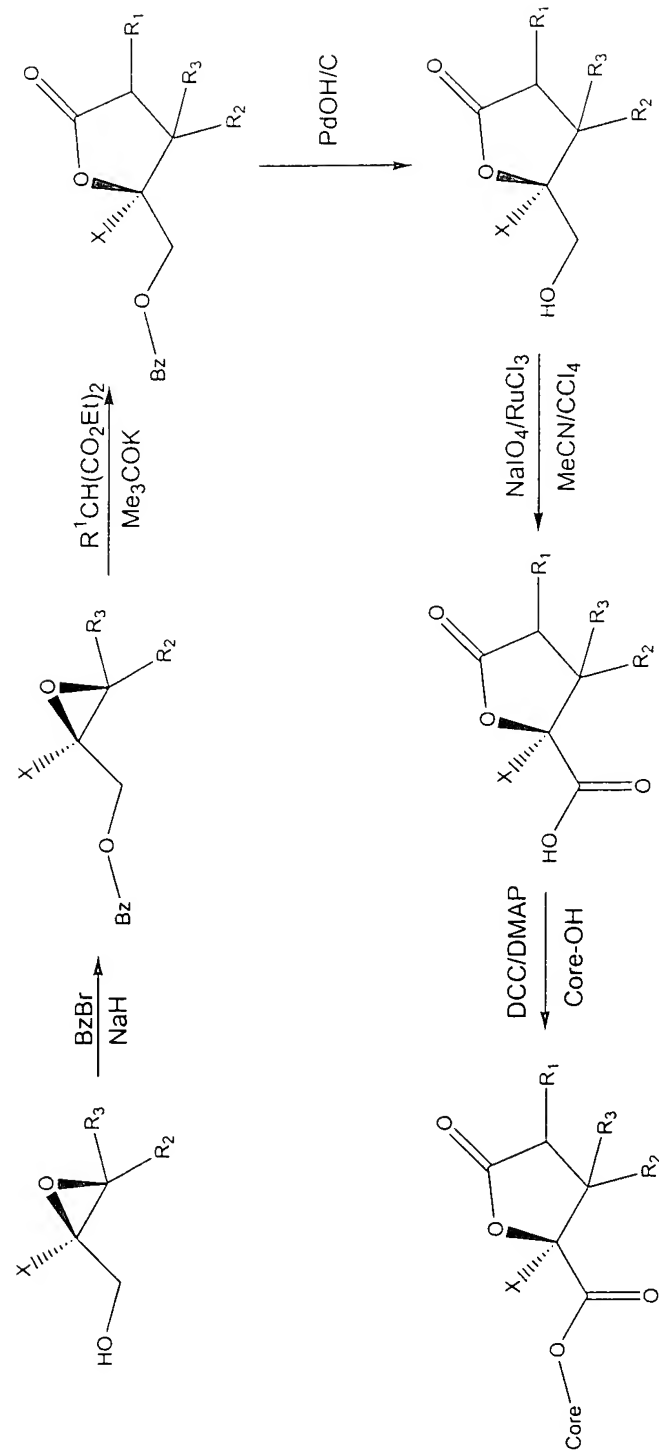


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Scheme 2

# Synthesis of Lactone Ester



[illegible]

Table I

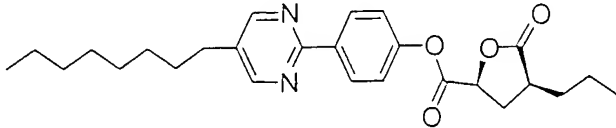
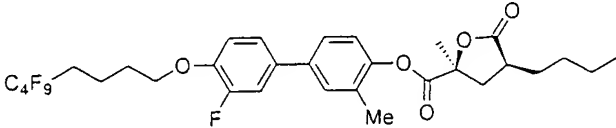
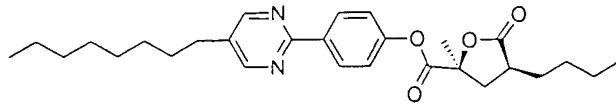
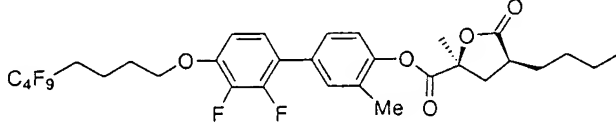
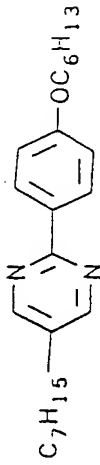
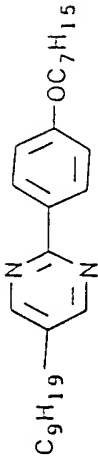
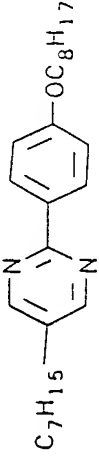
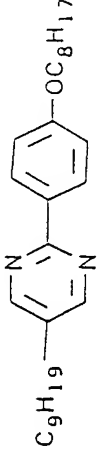
MDW	Structure	Ps (nC/cm <sup>2</sup> )
1198		40.1
1201		21.1
1212		11.1
1215		15

Table 2 Composition of MX 6111

MX 6111	
Structure and MDW #	% Composition
<p>MDW 1</p>  <chem>CCCCCc1ccncc1C1=CC=C(C=C1)OCCCC</chem>	5.6
<p>MDW 2</p>  <chem>CCCCCCCCc1ccncc1C1=CC=C(C=C1)OCCCCC</chem>	5.6



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DATE 01-14-2009 BY 60322 UCBAW/DK/STP  
REASON: THIS DOCUMENT DOES NOT CONTAIN  
INFORMATION THAT IS UNCLASSIFIED

<p>MDW 3</p>  <chem>CCCCCCCCc1ccc(C2=CC=CC=N2)cc1</chem>	<p>5.6</p>
<p>MDW 4</p>  <chem>CCCCCCCCCc1ccc(C2=CC=CC=N2)cc1</chem>	<p>7.2</p>



1. The following information is being provided for your information only. It is not intended to be used as a basis for any legal action. The information is being provided for your information only. It is not intended to be used as a basis for any legal action.

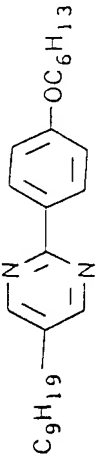
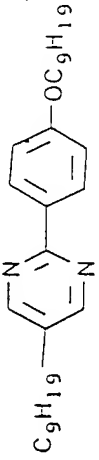
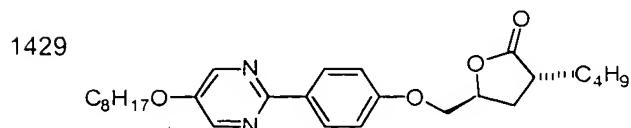
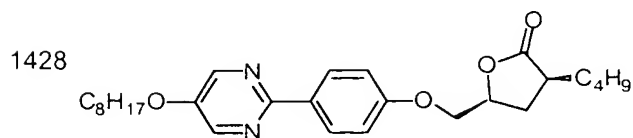
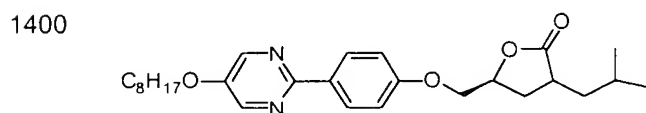
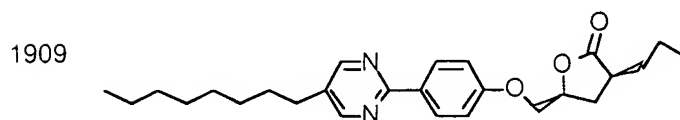
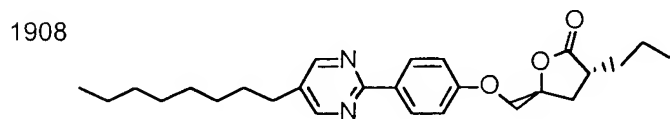
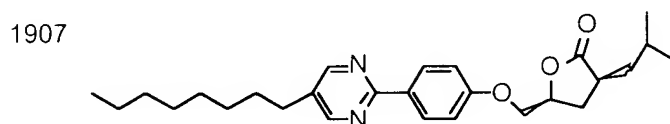
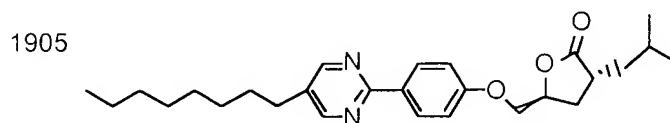
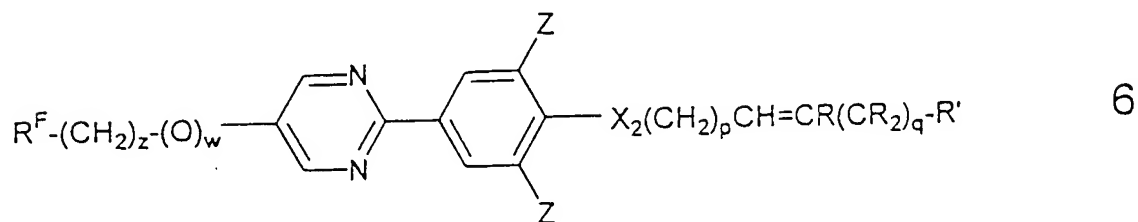
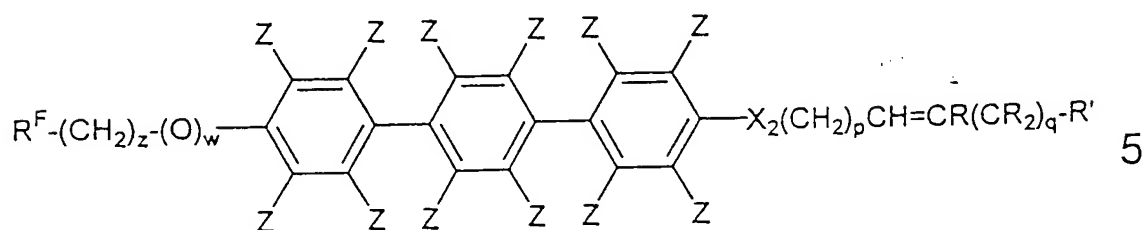
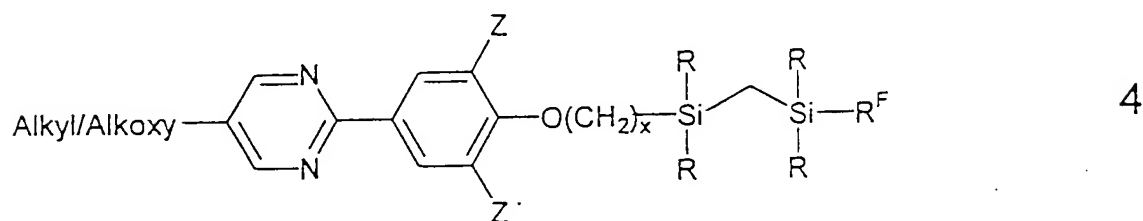
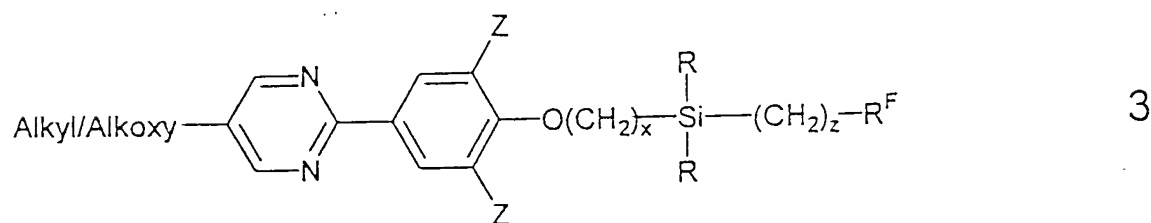
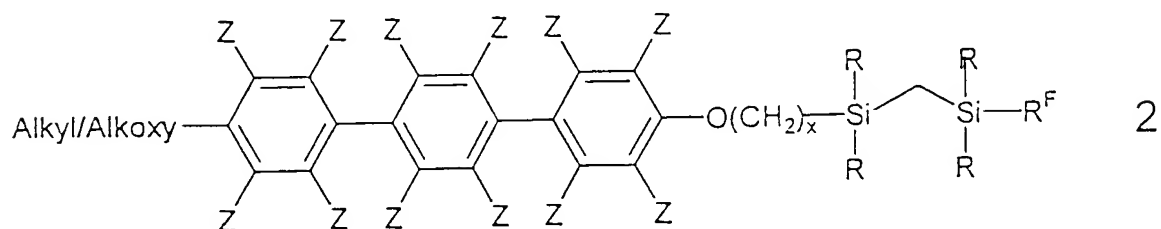
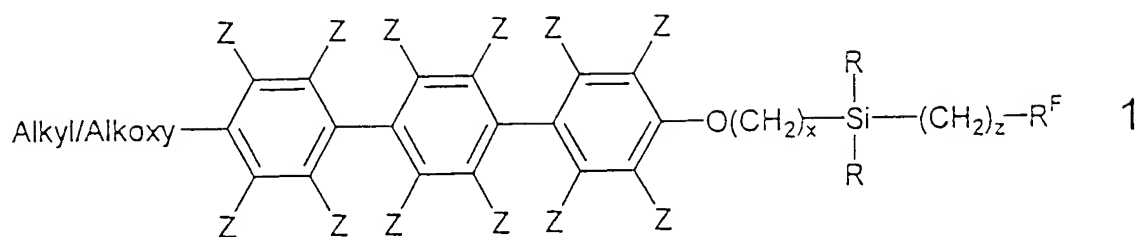
MDW 764		9.6
MDW 1287		33.6

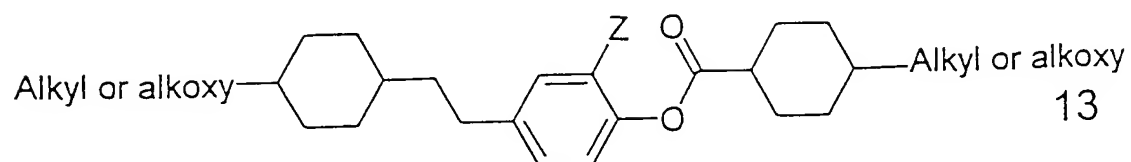
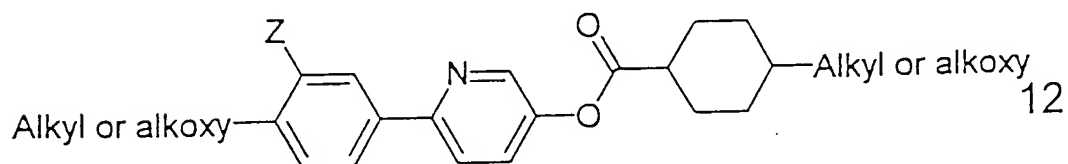
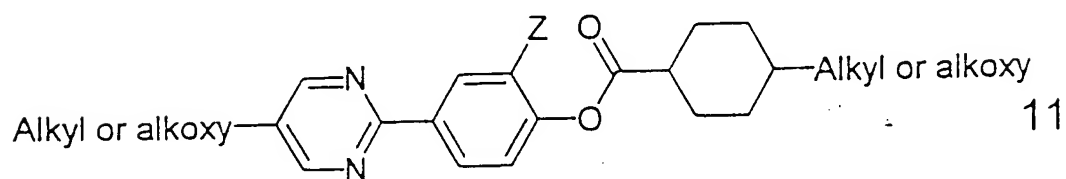
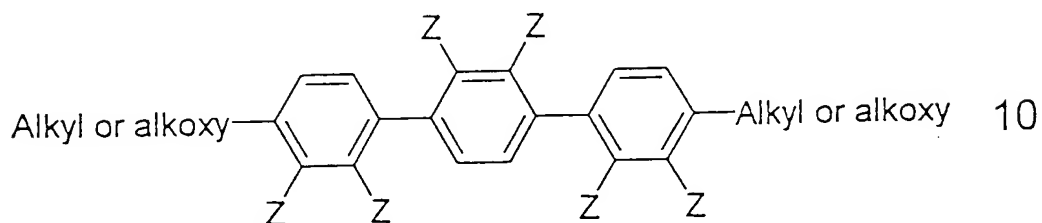
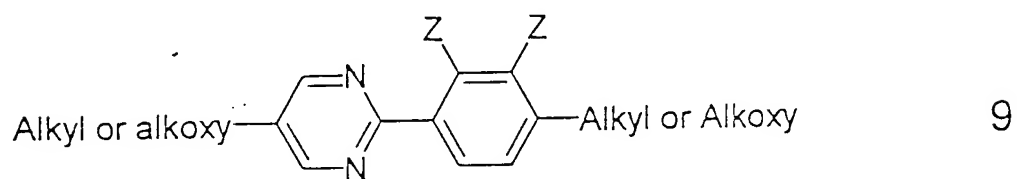
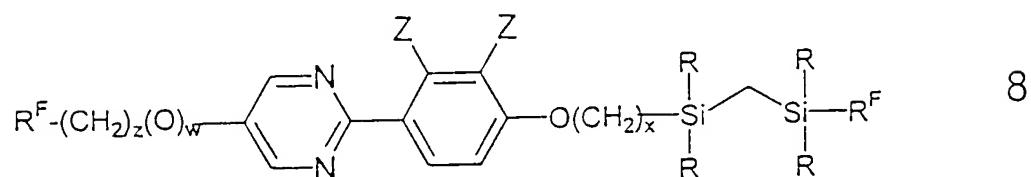
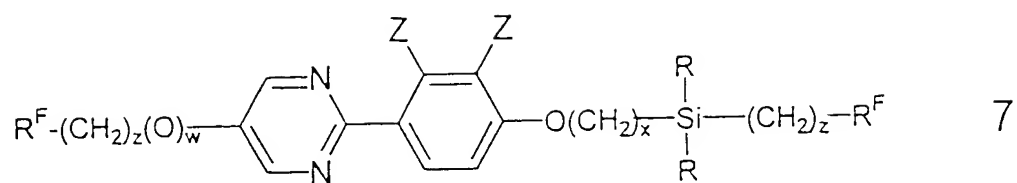
Table 3

lactone ether comparison compounds				
5% mix in MX6111				
	MIXES	Optical Phase	Ps	Elec-Rise Time
MDW #				
	10083	I 73.5 N 69 A 50.4 C <25 X	12.9	125
1400		X 133 I // I 115 X		
	10084	I 73.5 N 68.5 52 C <25 X	12.4	125
1428		X 133 I // I 115.1 X		
	10085	I 73.5 N 68.7 A 53.3 C <25 X	5.9	250
1429		X 112.4 I // I 99.1 X		
	10104	I (72.8-68.5) N 65.7 A 48 C <25 X	15.4	85
1905		X 100 I // I 78.9 X		
	10105	I (70.7-65) N 63 SmX 51.5 C < 25X	7.4	165
1907		OIL		
	10106	I (74.5-68.1) N 67.5 A 50 C <25 X	14.3	120
1908		X 98 I // I 108 X		
	10107	I (71-62) N 51 SmX 48 C <25 X	3.5	357
1909		X 90 I // I 25 X		

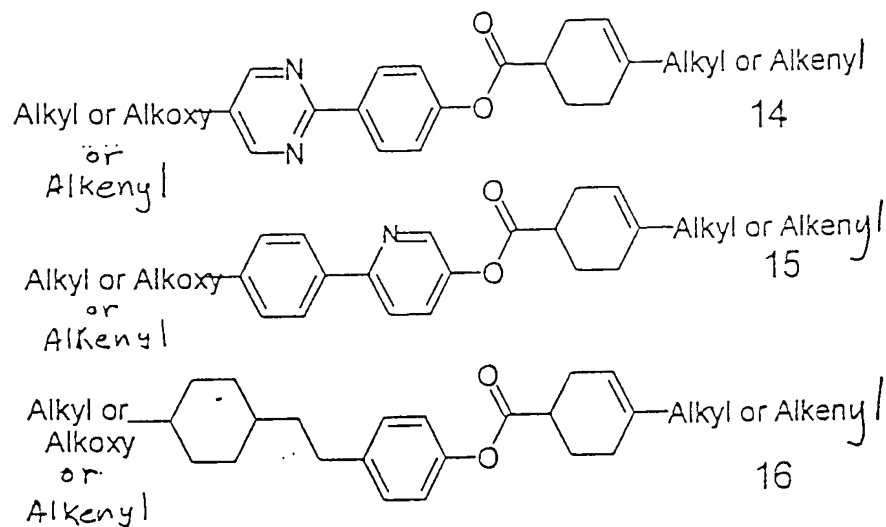




Scheme 4 (Page 2)

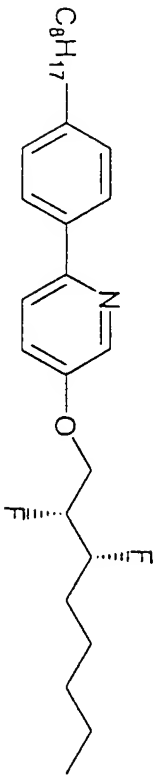
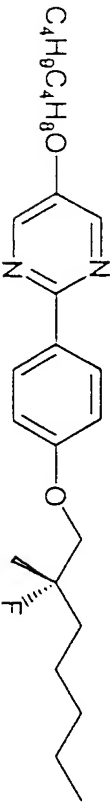
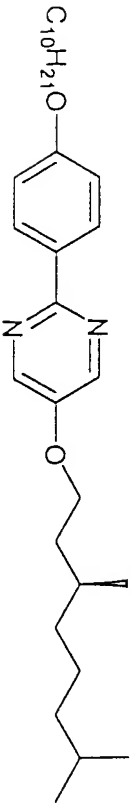
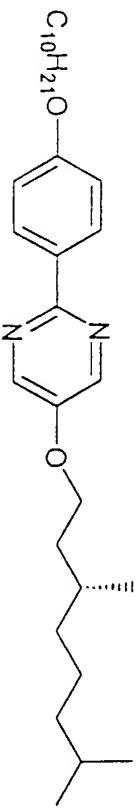
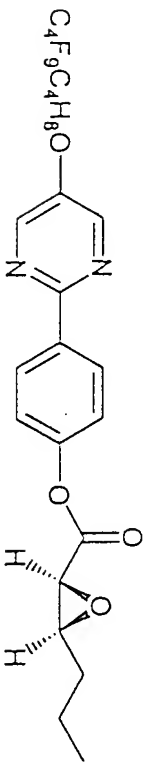


Scheme 4 (Page 3)



wherein  $p$ ,  $x$  and  $z$  are integers ranging from 1 to 20, inclusive,  $q$  is 0 or an integer ranging from 1 to 20, inclusive;  $w$  is 0 or 1;  $R$  are alkyl groups, preferably having from 1 to 6 carbon atoms;  $R'$  is an alkyl group having from 5 to 20 carbon atoms;  $R^F$  is a perfluoroalkyl group;  $Z$  is H or a F; and alkyl or alkoxy groups are those that have 1 to 20 carbon atoms.

Scheme 5 (Page 1)

MDW #	Structure	Phase Diagram
950		X <-90 - I - 94->
987		X <-21----- SmC* <-54- SmA <-63-I -53-> S? -57->
644		X <-20- N <-41 - I -43-> - 47->
699		
139		X - 75-> I <-86-

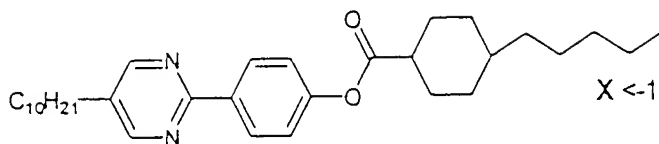


MDW #

Structure

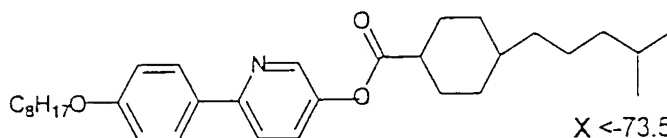
Phase Diagram

337

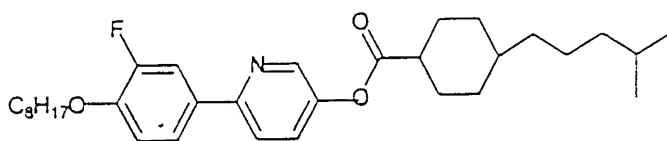


X &lt; -100- C &lt; -105- N &lt; -169- I

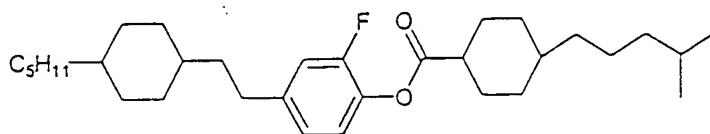
1135

X < -73.5 -S? < -85- C < -104 - A < -175- N  
< -186- I

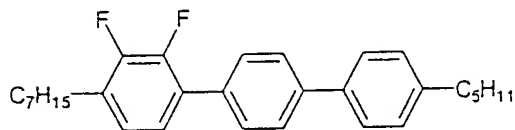
1638



1458

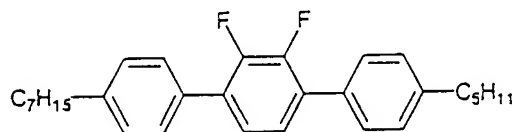


1671

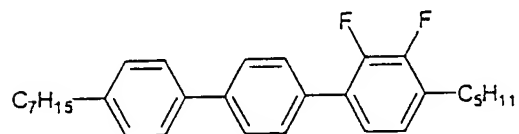


X -56-&gt; C -106-&gt; A -131-&gt; N -136-&gt; I

1673

X -37-> N -112-> I  
X < -24 C

1674



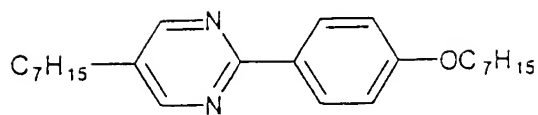
X -66-&gt; SI -75-&gt; C -119-&gt; A -135-&gt; N -137-&gt; I

MDW#

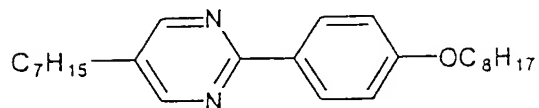
Structure

Phase Diagram

31

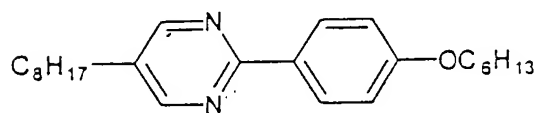


3

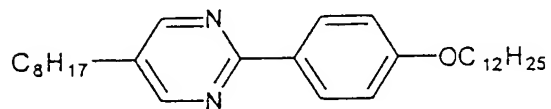


X -49-&gt; A -44-&gt; N -69.5-&gt; I

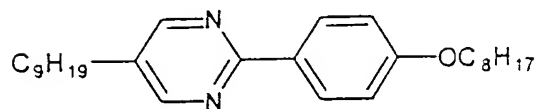
1695



5

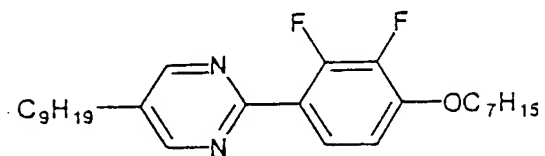
X -43.2->C -62.4->A -66.8->N  
-68.2-> I

4

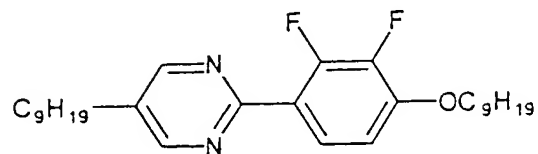


X -33-&gt;C -60-&gt;A -74.5-&gt; I

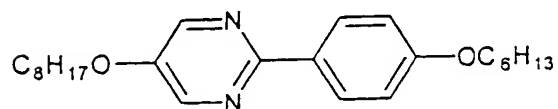
913

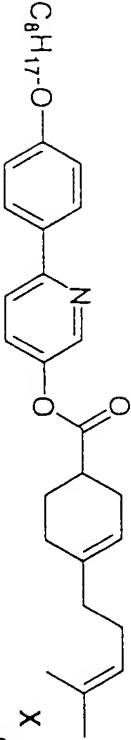
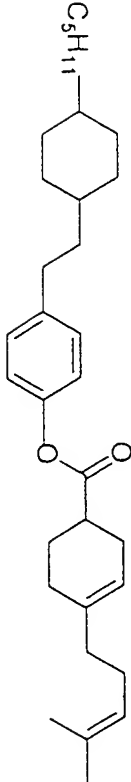
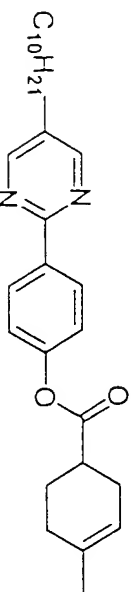
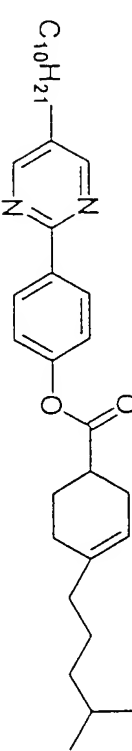
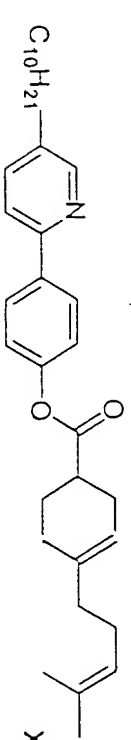
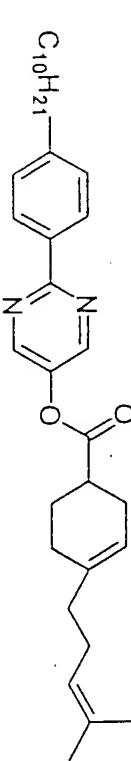
X -43->C -50-> I  
<-44- <52-

911

X -44->C -52-> I  
<-37- <-52-

374

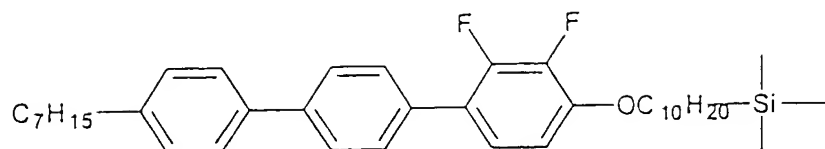


MDW	Structure	Phase Diagram
1054		X <----- C <-135- N<-150- I -55-> Sx -82->
942		
576		X <-35- S? <-45- C <-68- N<-107- I -50-> -54->
1059		
336		X <-27- C <-83- N<-106- I -40->
577		

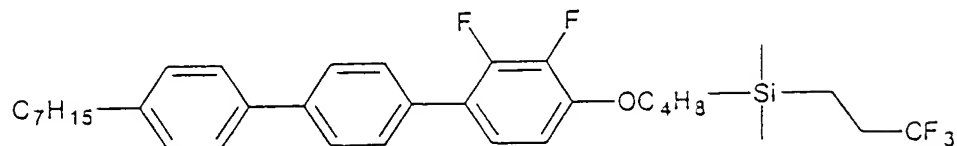
MDW#

Structure

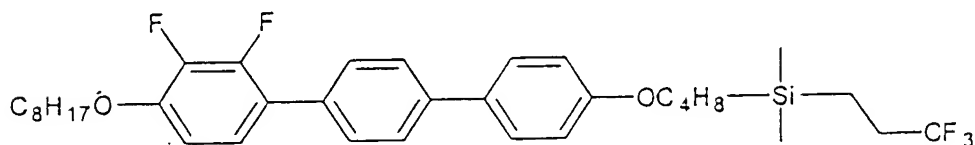
1701



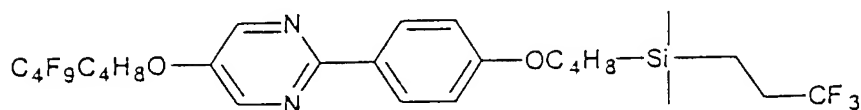
1669



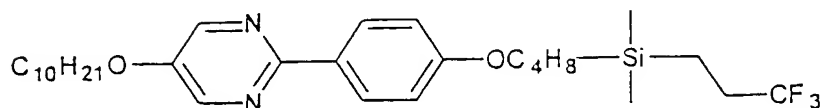
1658



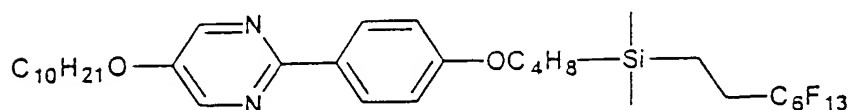
1592



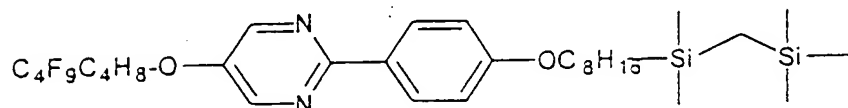
1532



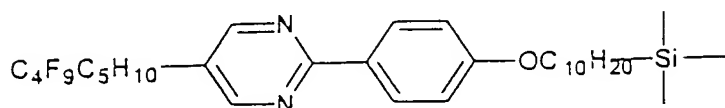
1632

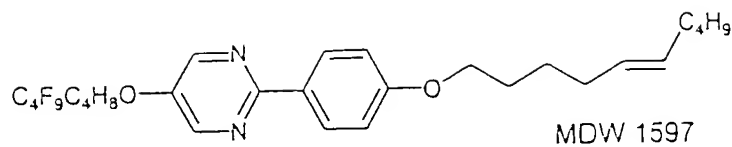


1586

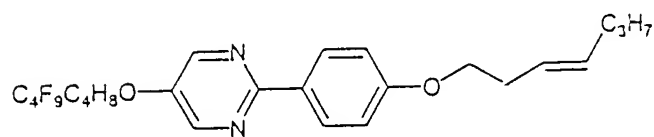


1709

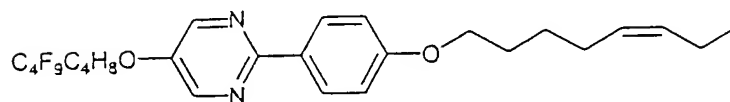




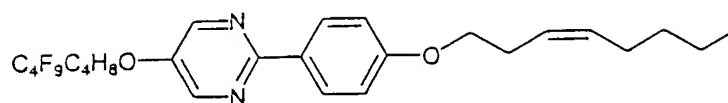
Cr 64.9 SmC 100.4 SmA 102.4 I  
43.3 99.6 101.0



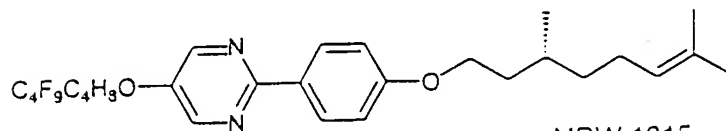
Cr 61.7 SmC 135.0 I  
57.7 134.6



Cr 70.7 SmC 113.8 SmA 115.4 I  
60.7 113.8 114.6



Cr 59.0 SmC 114.2 SmA 121.0 I



Cr 62 SmA 67 I

